## Bio-inspired/Water activated Systems for CO2 capture

Gonçalo V. S. M. Carrera, Luís C. Branco, Manuel Nunes da Ponte

### 1<sup>st</sup> Workshop CO2 SEQUESTRATION AND UTILIZATION at REQUIMTE





#### Introduction

- In the present environmental and energetic context frameworks for CO<sub>2</sub> capture are highly sought.
- Commercial systems in the market available for more than 60 years – Aqueous solutions of alkanolamines
- Major drawbacks:
  - Requirement of dilution of the capture agent in water (in order to avoid corrosion and mitigate excessive release of heat during reaction)
  - Moderate performances in CO<sub>2</sub> capture (7 wt% of CO<sub>2</sub> uptake in 30% aqueous solution of ethanolamine)
  - High energy demand for CO<sub>2</sub> stripping, due to the high heat capacity of water.
  - 4 Additionally, the solvent is lost during operations

Is here presented alternative CO<sub>2</sub> capture systems based on compounds from chiral pool, cheap carboxylic acids and organic superbases

#### Reversible Systems Based on CO<sub>2</sub>, Amino-acids and Organic Superbases

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Organic superbases

Aminoacids from chiral pool

#### Main Results

Compound	$T_d$
	(°C)
[TMGH <sup>+</sup> ] <sub>2</sub> [GlyCOO <sup>-</sup> ]	> 120
[DBUH <sup>+</sup> ] <sub>2</sub> [GlyCOO <sup>-</sup> ]	106.36
[TMGH <sup>+</sup> ], [AlaCOO <sup>-</sup> ]	> 88
[DBUH <sup>+</sup> ], [AlaCOO <sup>-</sup> ]	101.77
[TMGH <sup>+</sup> ] <sub>2</sub> [ValCOO <sup>-</sup> ]	> 120
[DBUH <sup>+</sup> ] <sub>2</sub> [ValCOO <sup>-</sup> ]	96.35
[TMGH <sup>+</sup> ] <sub>2</sub> [LeuCOO <sup>-</sup> ]	109.07
[DBUH <sup>+</sup> ] <sub>2</sub> [LeuCOO <sup>-</sup> ]	95.6
[TMGH <sup>+</sup> ] <sub>2</sub> [PheCOO <sup>-</sup> ]	93.79
[DBUH <sup>+</sup> ] <sub>2</sub> [PheCOO <sup>-</sup> ]	86.91
[TMGH <sup>+</sup> ] <sub>2</sub> [TrpCOO <sup>-</sup> ]	> 74
[DBUH <sup>+</sup> ] <sub>2</sub> [TrpCOO <sup>-</sup> ]	79.62

 Amino-acid based carbamate salts were successfully prepared using CO<sub>2</sub> and an Organic Superbase.

For DBU based salts, as the size of R- group of the amino-acid increases the value of T<sub>d</sub>, associated to CO<sub>2</sub> release, decreases.

Saccharides and derived structures

$$\begin{bmatrix} R \\ H \end{bmatrix}_{X} \qquad R, R' = \begin{bmatrix} H & \text{or } C & \text{or } O \end{bmatrix}$$

Organic Superbases

SB

$$\begin{array}{c}
R \\
R'
\end{array}$$

$$\begin{array}{c}
CO_2 \\
R'
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CO_2 \\
R'
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$$\begin{array}{c}
CO_2 \\
R'$$

$$\begin{array}{c}
R''
\end{array}$$

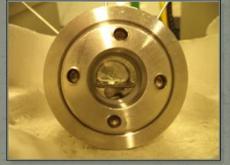
$$\begin{array}{c}
R''$$

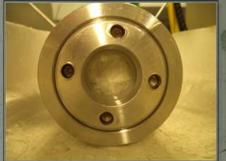
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CO, Capture systems based on saccharides and organic superbases

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- Best performance: D-Mannose:DBU  $(0.625:1 in equivalents) \rightarrow 13.9 wt\% of$ CO<sub>2</sub> uptake and 3.3/5 alcohol groups to carbonates.
- Energy Requirement for CO, strip  $\rightarrow$  (2790 kJ/kg CO<sub>2</sub>) Benchmark system  $\rightarrow$  (3873 kJ/kg CO<sub>2</sub>).

#### Aqueous Carboxylic Acid-Based Solutions for CO<sub>2</sub> Capture

In progress...

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